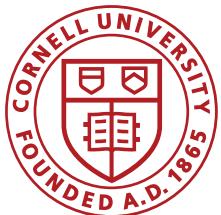


Anaerobic Fermentation to Produce Carboxylic Acids and Inactivate *Ascaris* Eggs

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Janice Liotta, Research Support Specialist
Dr. Dwight Bowman, PhD
Dr. Lars Angenent, PhD

Cornell University



Our Goal:
**Inactivate pathogens in fecal
solids through *in-situ* production
of carboxylic acids**

Outline

Introduction

- What are carboxylic acids?
- Carboxylic acids and pathogens
- Production of carboxylic acids

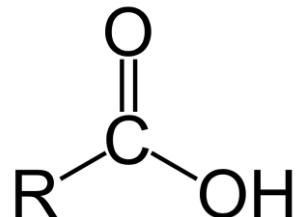
Methods and Results

- Batch fermentation
- Inactivation of *Ascaris* eggs

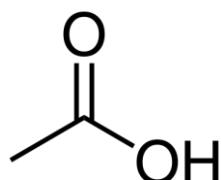
Summary and Future Work

What is a carboxylic acid?

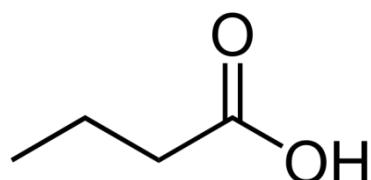
Carboxylic acid: a weak organic acid containing a carboxyl group



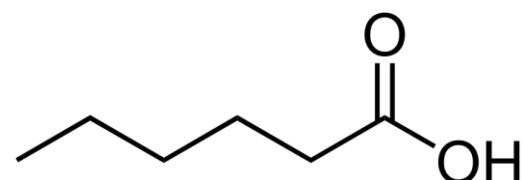
We most often think of *fatty acids* (hydrocarbon chains followed by a carboxyl group).



Acetic acid
C2

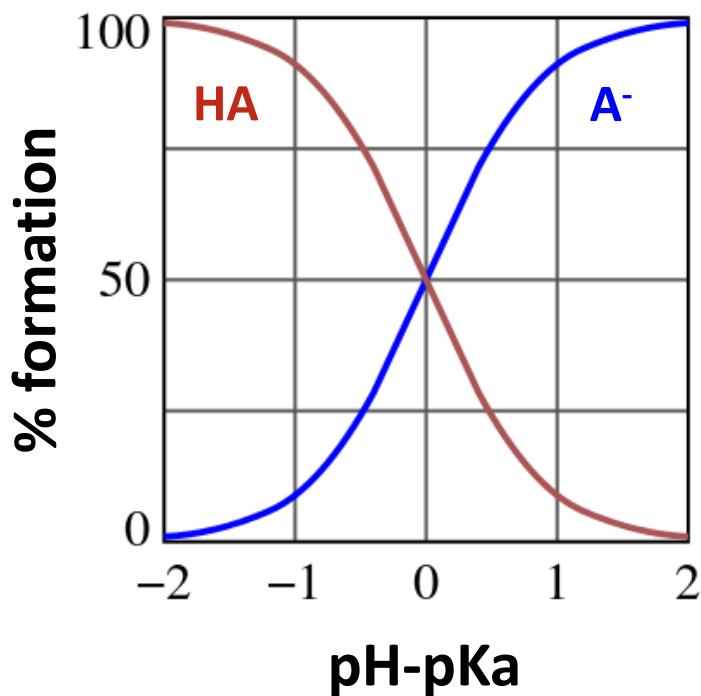
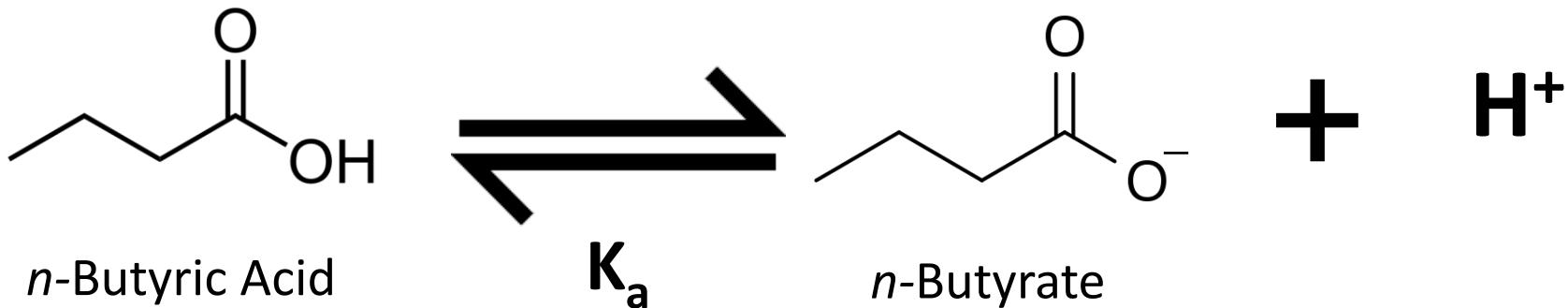


n-Butyric acid
C4



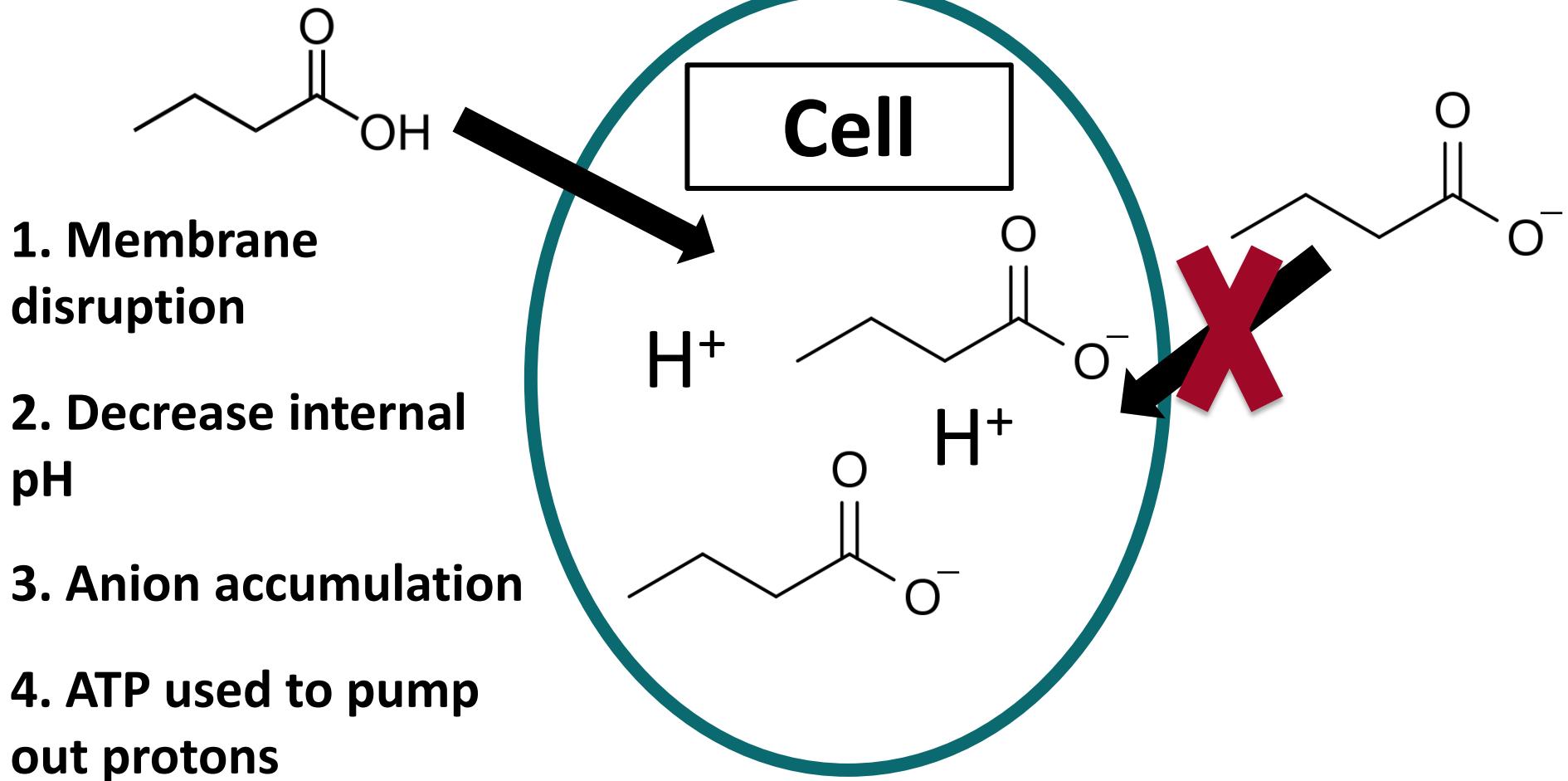
n-Caproic acid
(Hexanoic Acid)
C6

A note about weak acids



$$pH = pK_a + \log\left(\frac{[A^-]}{[HA]}\right)$$
$$pK_a = 4.8$$

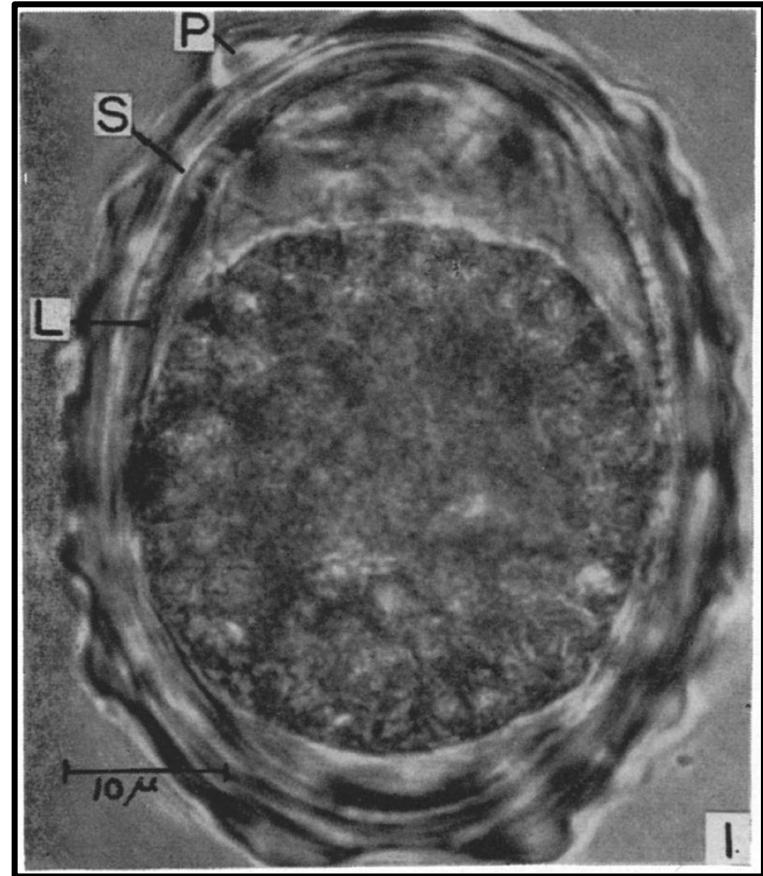
Carboxylic acids and pathogens



Ascaris as a model pathogen



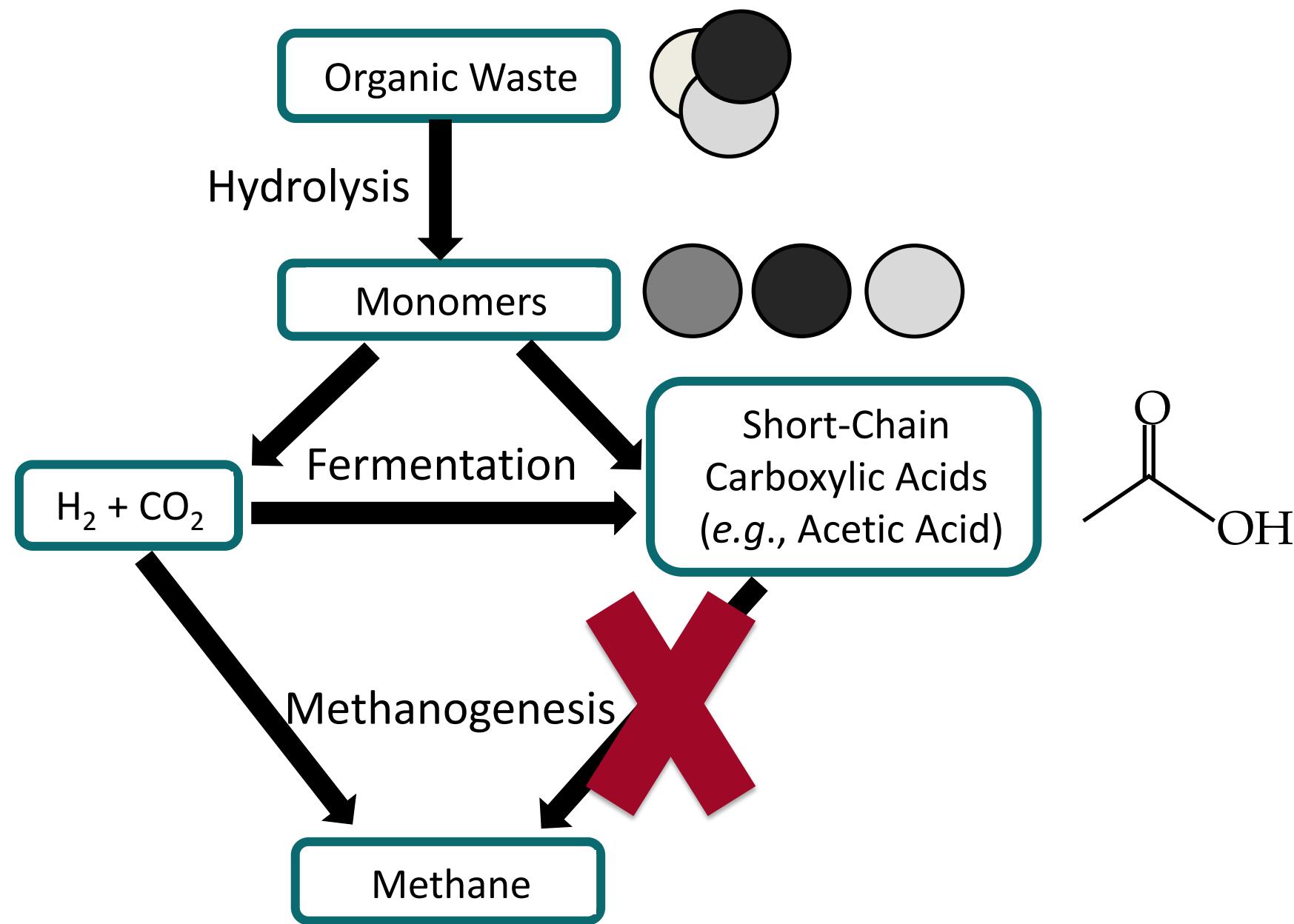
medicalstate.tumblr.com



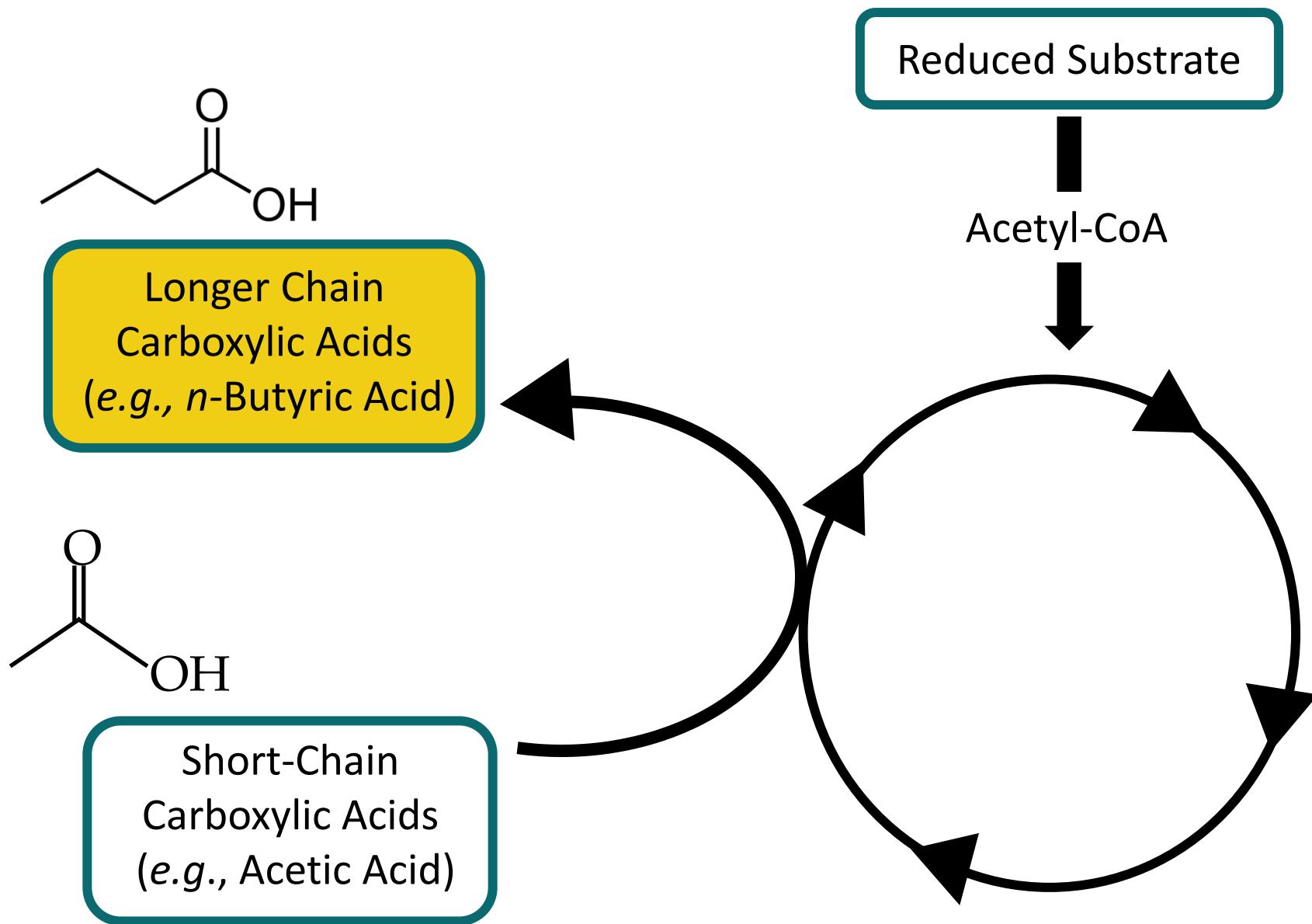
Rogers, 1956

Butkus, M. A., et al. 2011. Inactivation of *Ascaris suum* by short-chain fatty acids. *Applied and environmental microbiology*, 77(1), 363-366.

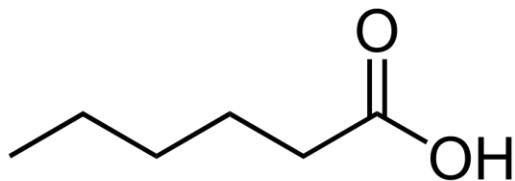
Carboxylic acid production and chain elongation



Chain Elongation



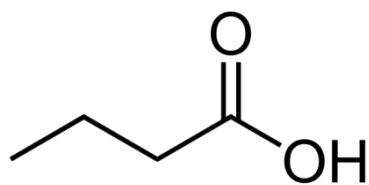
Chain Elongation



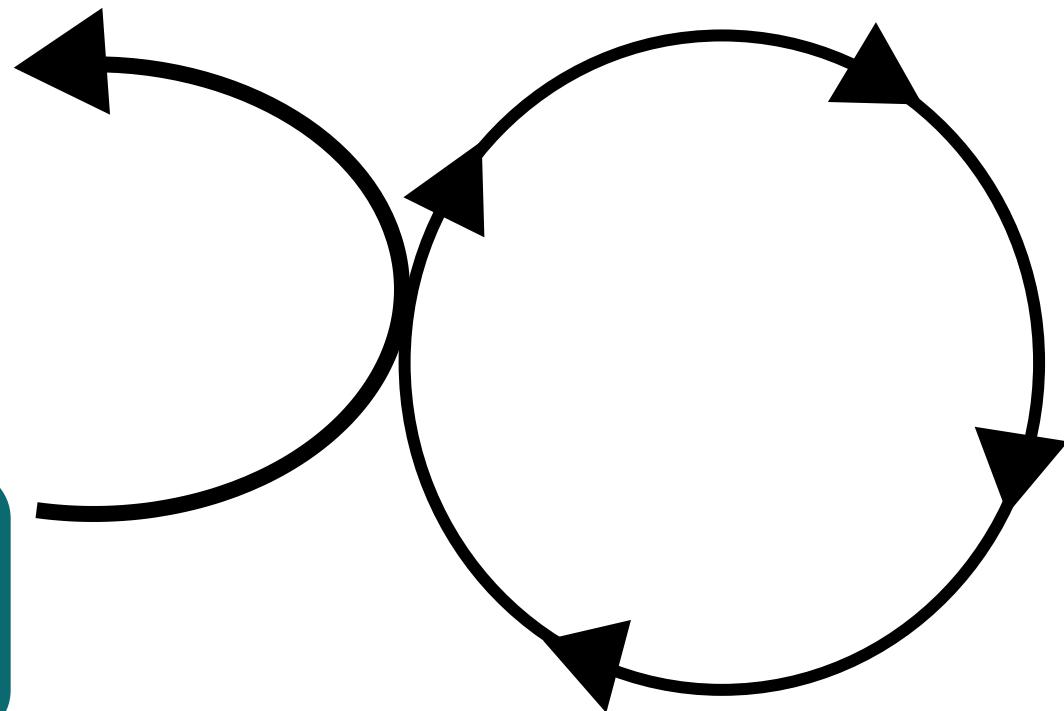
Reduced Substrate

Acetyl-CoA

Longer Chain
Carboxylic Acids
(e.g., *n*-Caproic Acid)



Short-Chain
Carboxylic Acids
(e.g., *n*-Butyric Acid)



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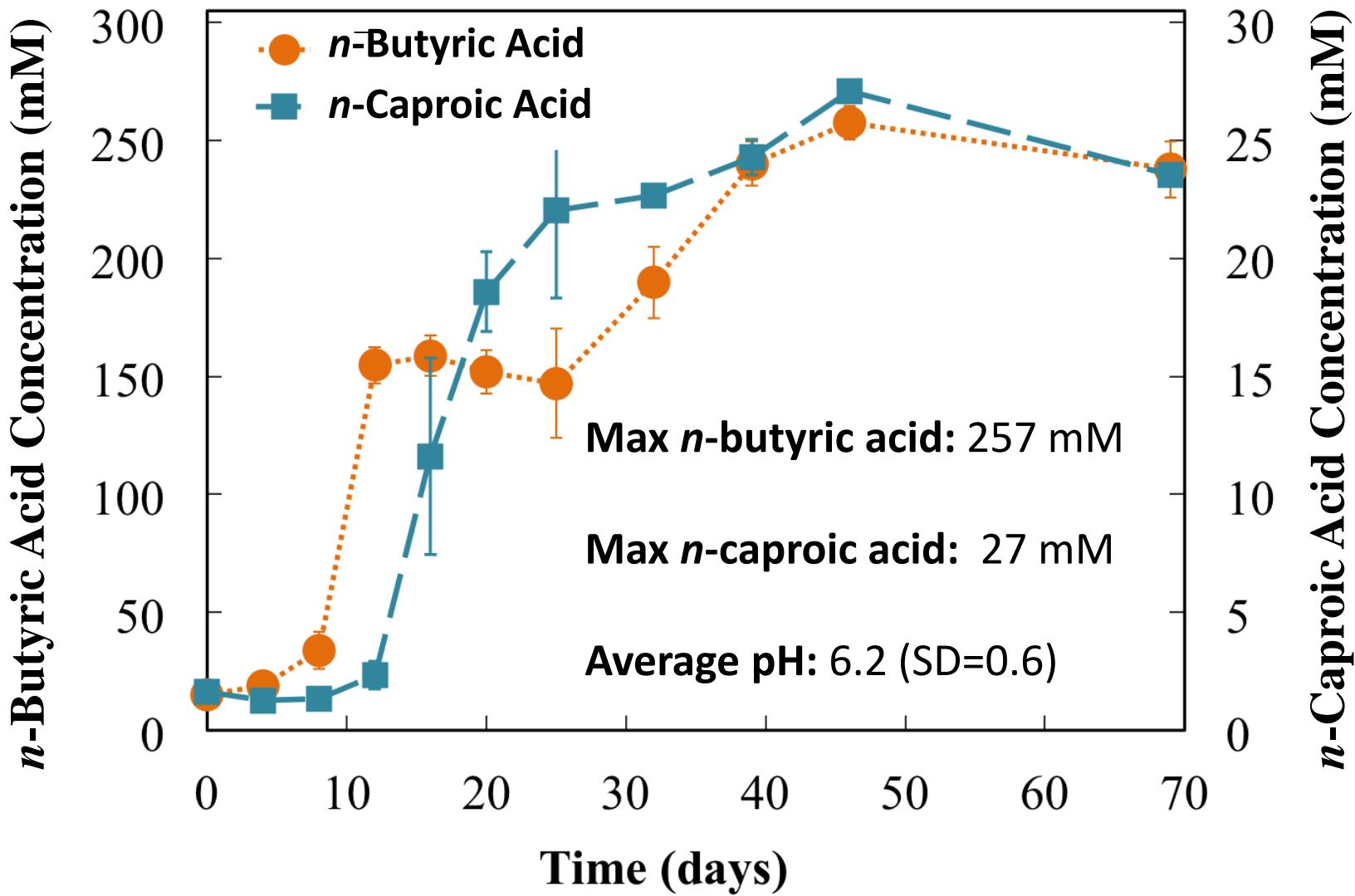
Summary and Future Work

Batch fermentation to produce carboxylic acids

- Fecal solids and inoculum from a carboxylate-producing reactor
- Incubated anaerobically at 30°C
- Sampled every 4-7 days for 69 days



Batch fermentation to produce carboxylic acids

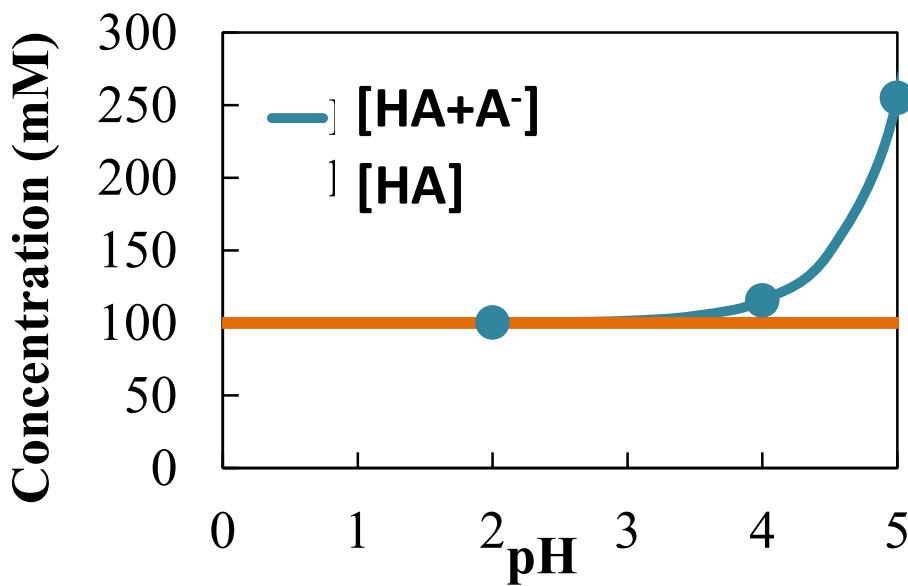


Parameters controlling *Ascaris* viability

- **Concentration**
- **pH**
- Carbon chain length
- Exposure time
- Temperature

Ascaris viability: Carboxylic acid concentration and pH

- Three pH levels tested ($\text{pH} = 2, 4, 5$)
- Same concentrations of uncharged acids (HA) tested at each pH

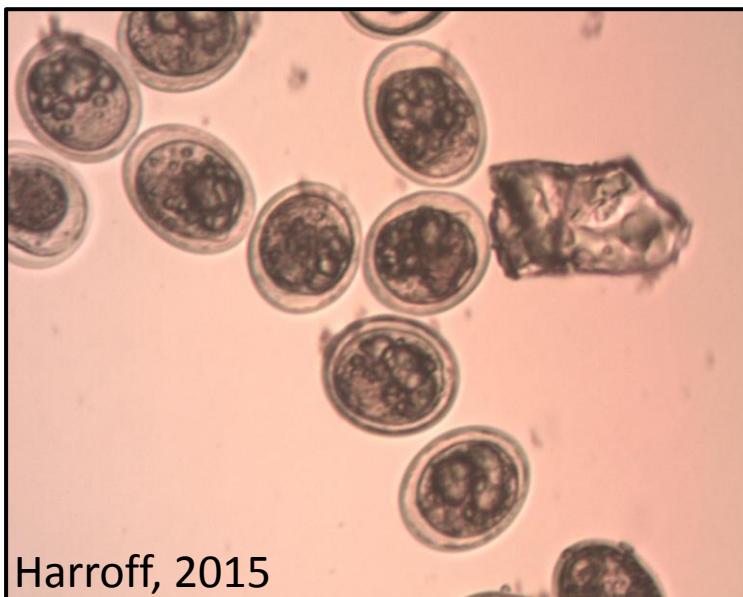


$$\text{pH} = \text{pK}_a + \log\left(\frac{[\text{A}^-]}{[\text{HA}]}\right)$$

$$[\text{HA} + \text{A}^-] = [\text{HA}] (10^{(\text{pH}-\text{pK}_a)} + 1)$$

Ascaris viability: Carboxylic acid concentration and pH

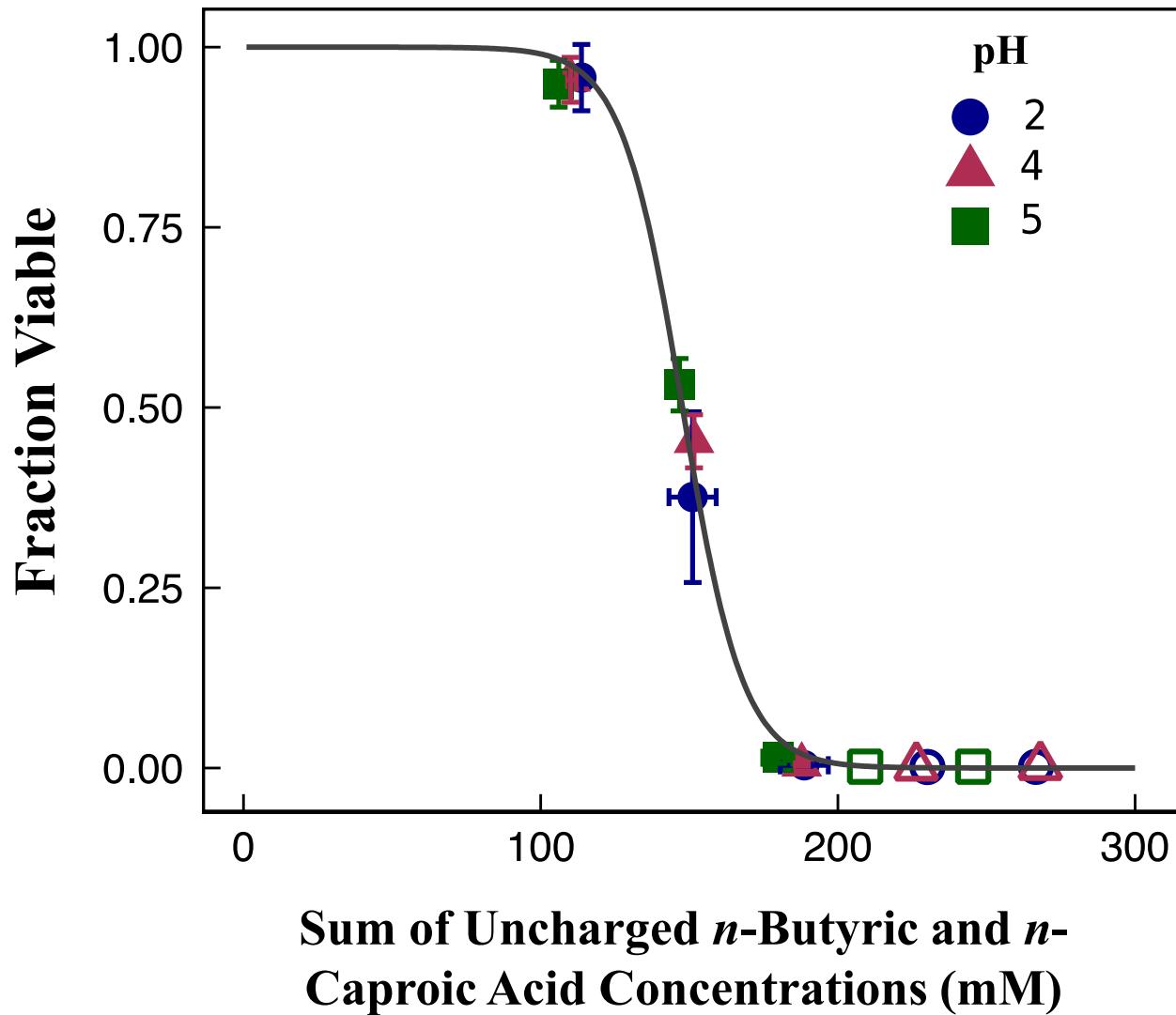
- Exposed for 3 days at 37°C
- Washed and incubated for 21 days at 30°C
- 500 eggs examined for viability



Harroff, 2015



Ascaris viability not dependent on anion concentration or pH



$$y = \frac{1}{1 + \exp(-a(x - b))}$$

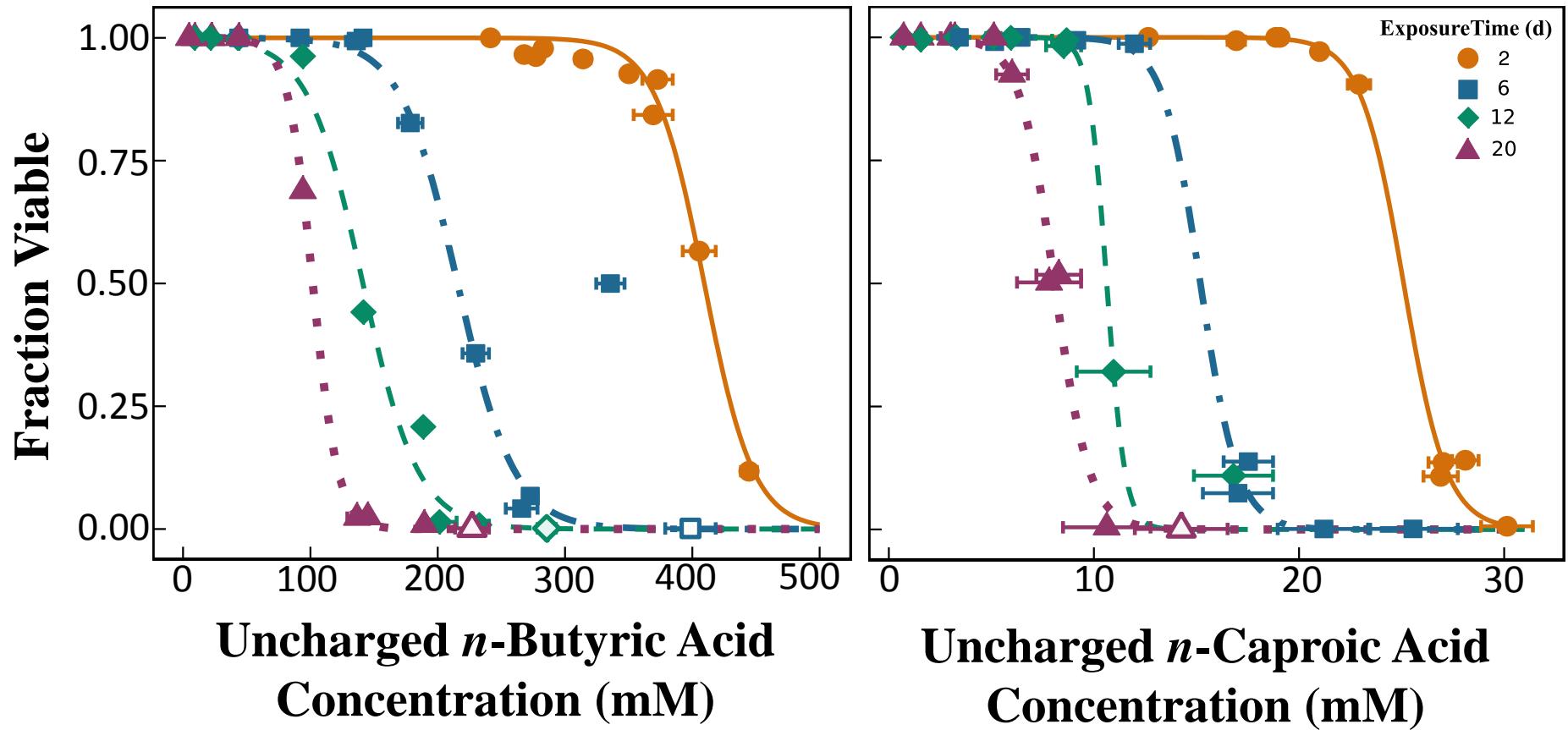
Parameters controlling *Ascaris* viability

- **Concentration**
- pH
- **Carbon chain length**
- **Exposure time**
- Temperature

***Ascaris* viability: Concentration, chain length, and exposure time**

- *A. suum* eggs exposed to *n*-butyric acid and *n*-caproic acid individually
- pH 2
- Concentration and exposure time varied
- T=30°C
- Anaerobic

Ascaris viability dependent on concentration, chain length, and exposure time



$$y = \frac{1}{1 + \exp(-a(x - b))}$$

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Summary and Future Work

Conclusions

- We can produce carboxylic acids from fecal solids.
- The concentration of uncharged acid is important.
- For individual acids, we can predict *Ascaris* inactivation based on exposure time and concentration.

Future Work

- Decrease pH through fermentation
- Understand different parameters relating to inactivation:
 - Temperature
 - Mixtures of acids
 - Matrix
- Scale up

Acknowledgements

Angenent and Bowman Labs

Ascaris inactivation

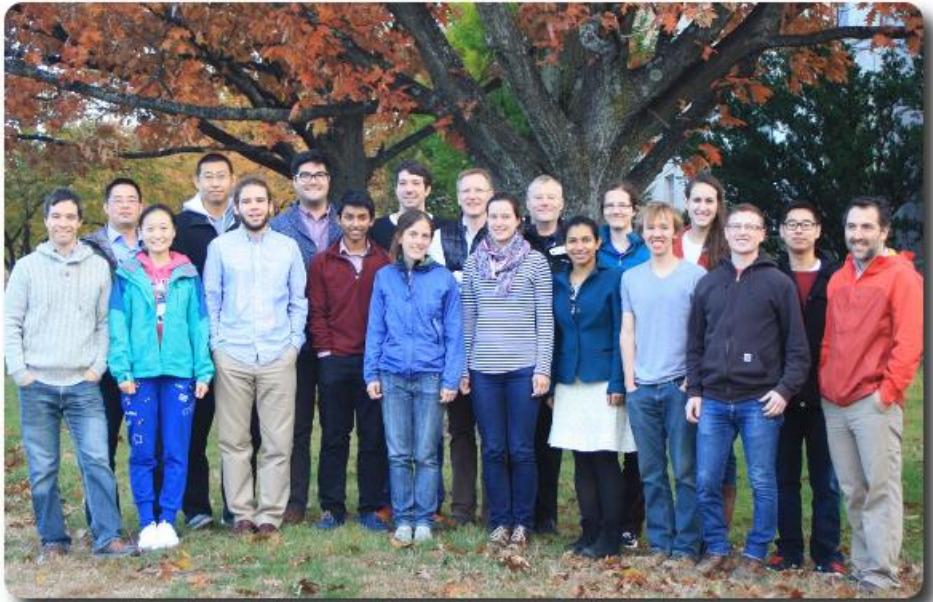
- Dr. Michael P. Labare, PhD
- Dr. Michael Butkus, PhD
- Dr. Nzuhat Islam, MS, DVM
- Zhu Dan, MEng

Statistics

- Dr. Lynn Johnson, PhD
Cornell Statistical Consulting Unit

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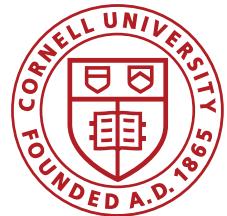


Atkinson Center
for a Sustainable Future

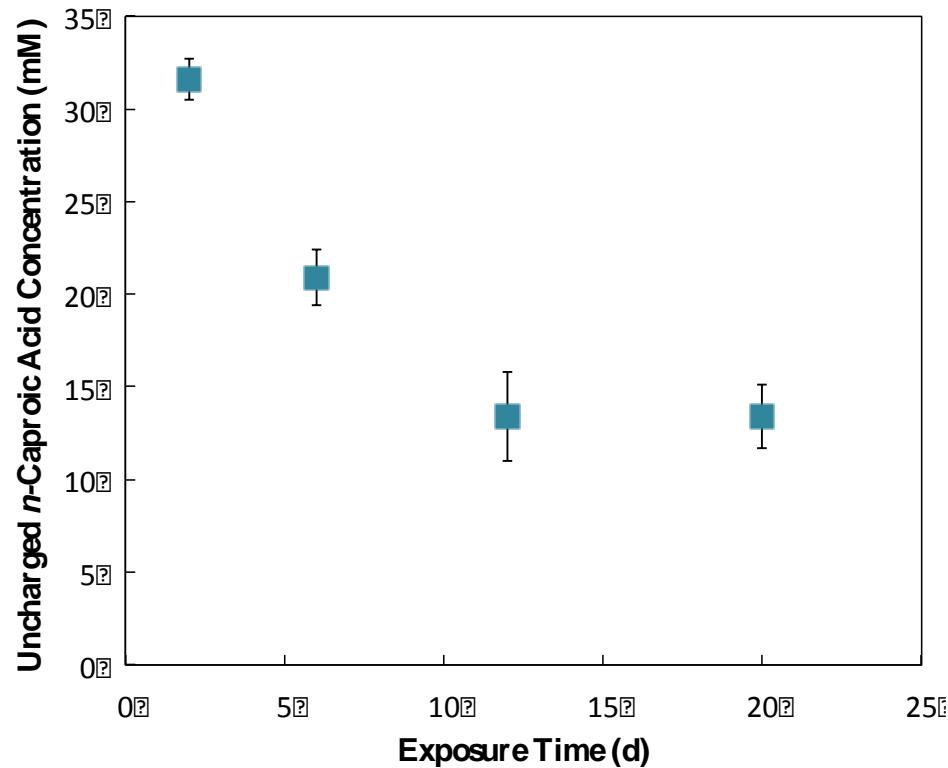
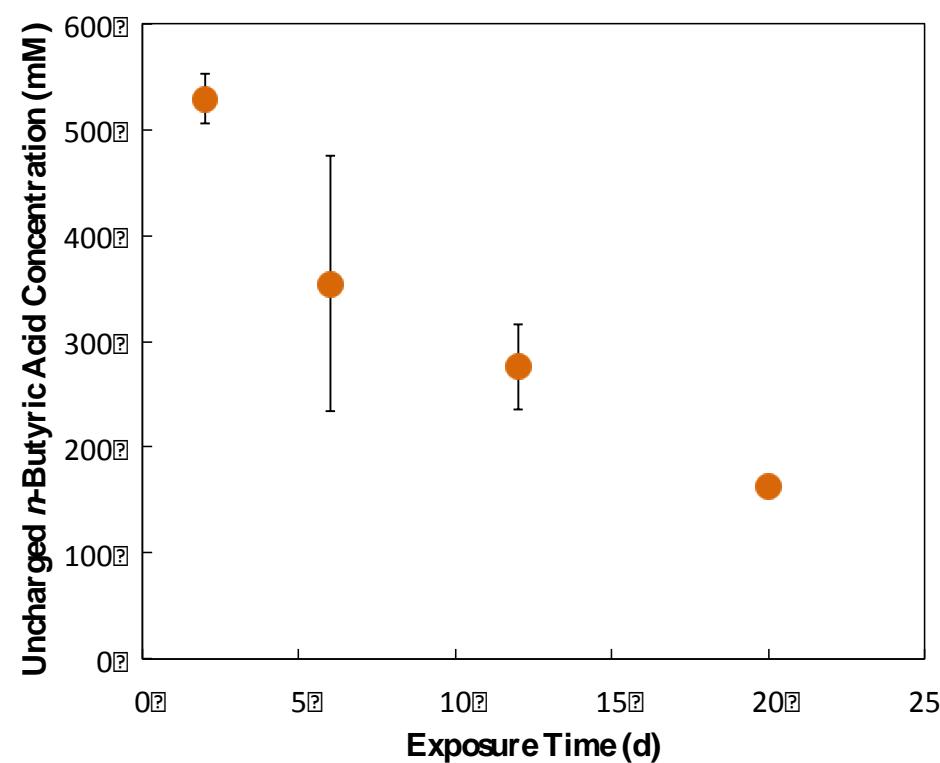


Conclusions

- We can produce carboxylic acids from fecal solids.
- The concentration of uncharged acid is important.
- For individual acids, we can predict *Ascaris* inactivation based on exposure time and concentration.



Predict time and concentration requirements for complete inactivation



Predict time and concentration requirements for complete inactivation

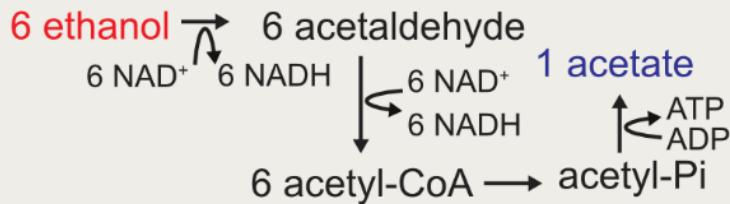
Acid	Exposure Time (d)	Predicted Uncharged Concentration Required to Reduce Viability Below Detection (0.2%) (mM)	95% Confidence Interval (mM)
<i>n</i> -Butyric	2	529	506-553
<i>n</i> -Butyric	6	354	234-475
<i>n</i> -Butyric	12	276	236-316
<i>n</i> -Butyric	20	163	158-169
<i>n</i> -Caproic	2	32	31-33
<i>n</i> -Caproic	6	21	19-22
<i>n</i> -Caproic	12	13	11-16
<i>n</i> -Caproic	20	13	12-15

Model parameters

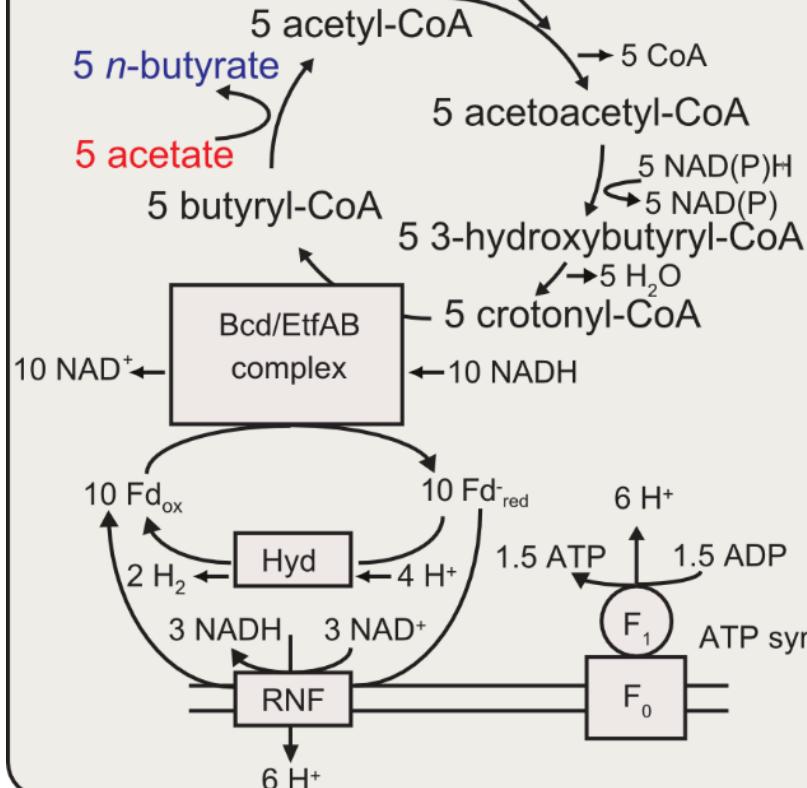
Acid	Exposure Time (d)	a (std error)	p-value a	b (std error)	p-value b	Mean Square Error	Mean Absolute Error
Butyric	2	-0.0519 (0.00480)	4.71*10^(-6)	409 (2.23)	8.75*10^(-16)	0.000981	0.0282
Butyric	6	-0.0448 (0.0203)	0.0582	216 (14.1)	3.23*10^(-7)	0.025	0.0641
Butyric	12	-0.0462 (0.00707)	0.000182	141 (3.85)	3.36*10^(-10)	0.00207	0.0303
Butyric	20	-0.0101 (0.00418)	9.26*10^(-9)	102 (0.388)	2*10^(-16)	2.58* 10^(-5)	0.00315
Caproic	2	-0.954 (0.0927)	6.87*10^(-6)	25.1 (0.200)	1.83*10^-14)	0.000987	0.0171
Caproic	6	-1.09 (0.199)	0.000585	15.2 (0.377)	1.58*10^(-10)	0.000676	0.0142
Caproic	12	-2.24 (1.11)	0.0784	10.6 (0.179)	7.22*10^(-12)	0.00121	0.013
Caproic	20	-1.17 (0.188)	0.000254	8.08 (0.103)	7.61*10^(-13)	0.00146	0.024

Reverse Beta-Oxidation

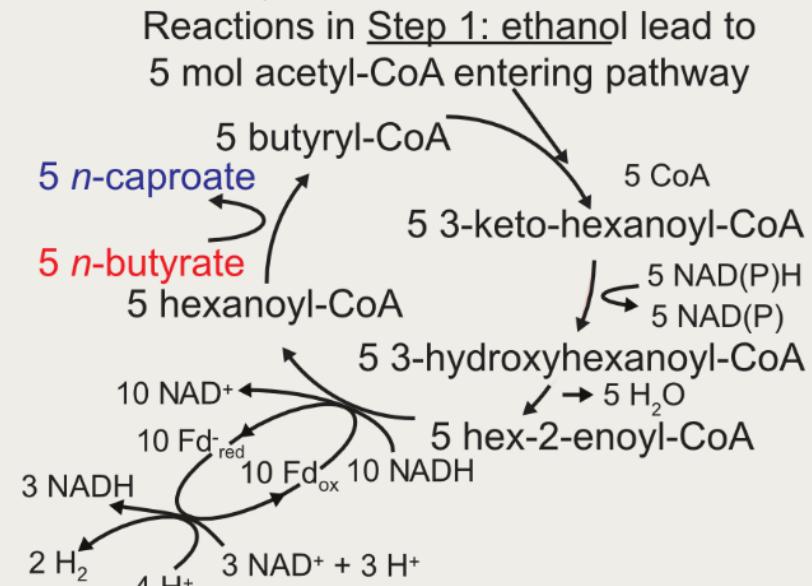
Step 1: ethanol



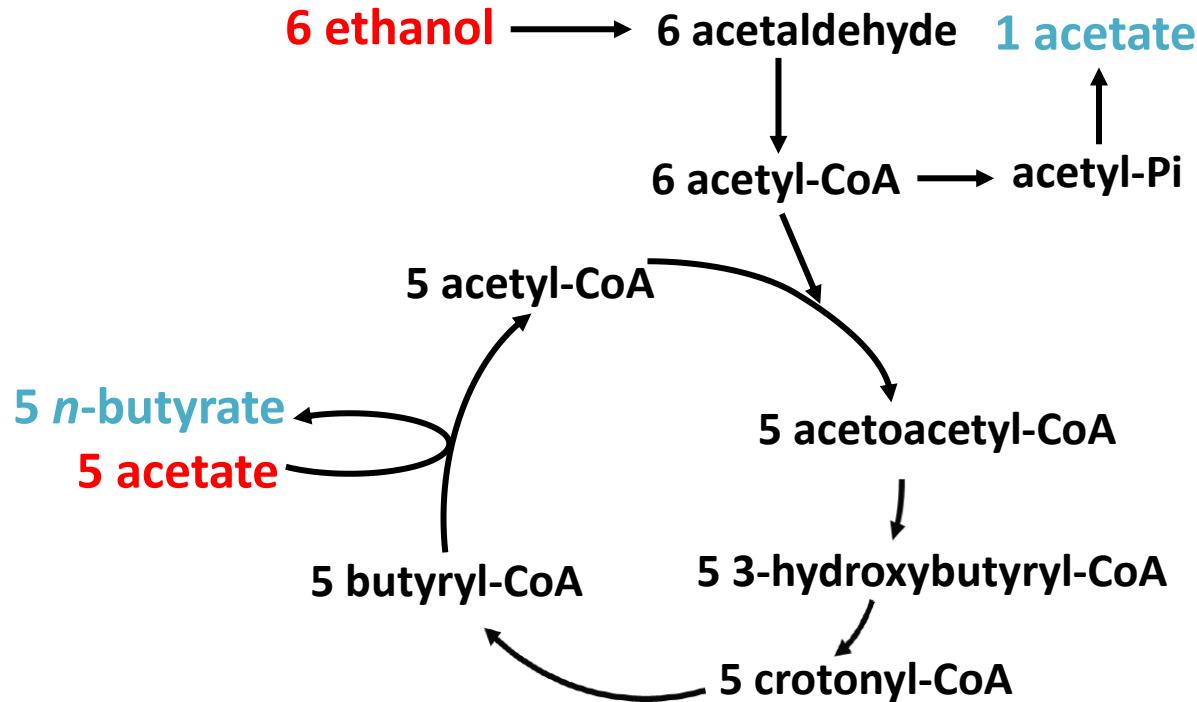
Step 2: acetate



Step 2: n-butyrate



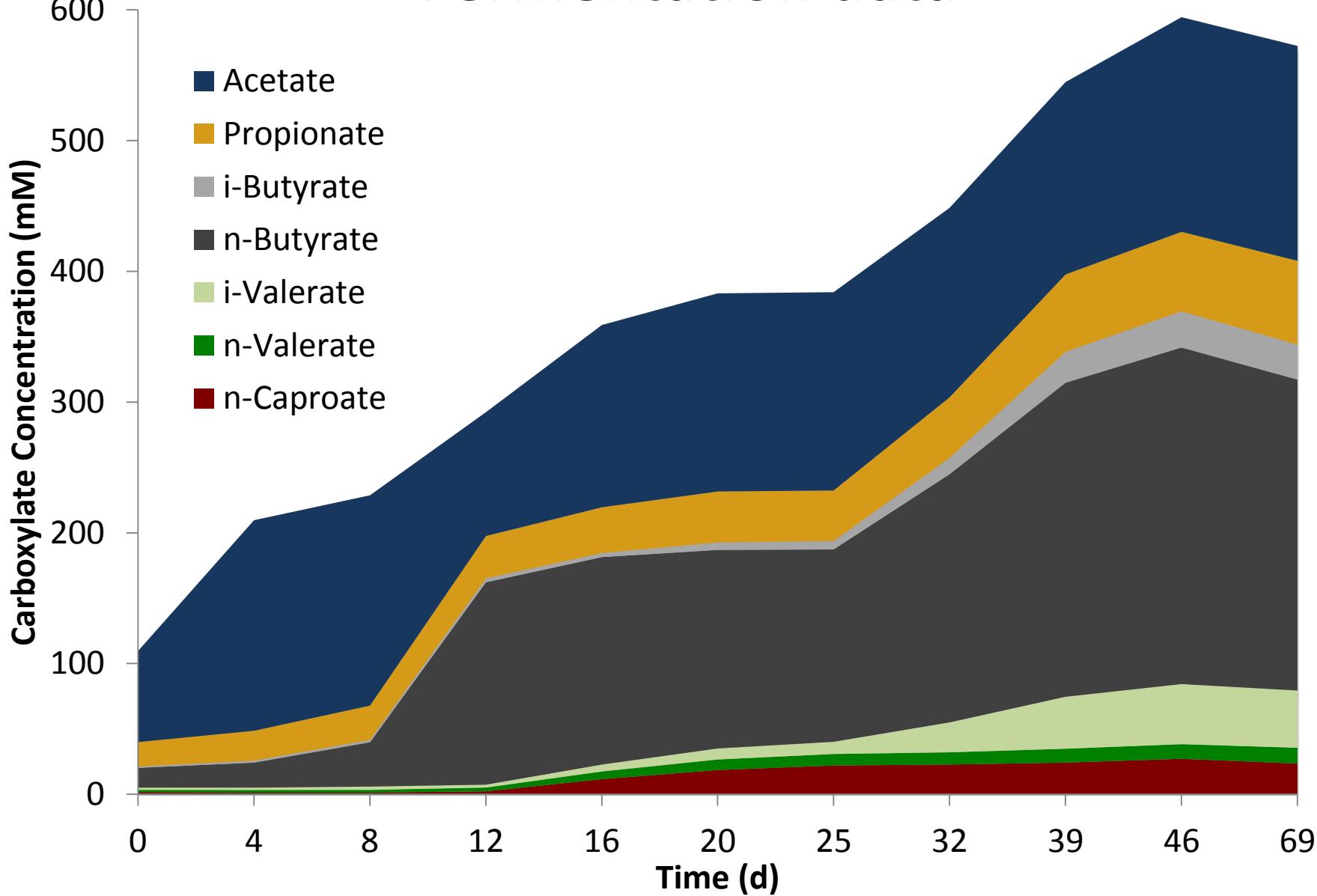
Reverse Beta-Oxidation: C2 to C4



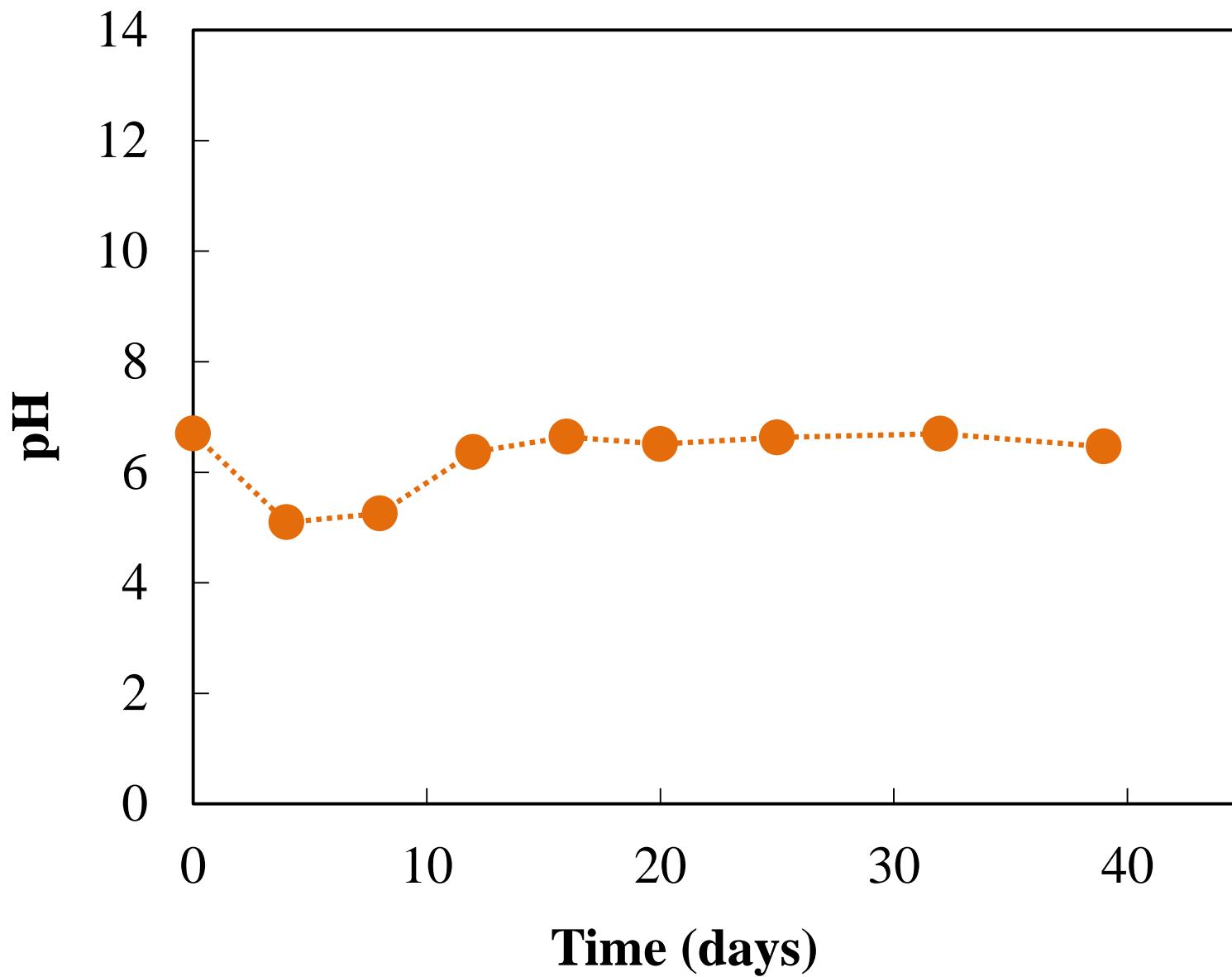
Concentrations for uncharged acid data

Treatment #	pH	Uncharged Butyric Acid (mM)	Uncharged Caproic Acid (mM)	Total Butyrate (mM)	Total Caproate (mM)
1	2	100	10	100	10.0
2	2	135	13.5	135	13.5
3	2	170	17	170	17.0
4	2	205	20.5	205	20.5
5	2	240	24	240	24.0
6	4	100	10	115	11.3
7	4	135	13.5	155	15.3
8	4	170	17	196	19.2
9	4	205	20.5	236	23.2
10	4	240	24	276	27.2
11	5	100	10	251	23.2
12	5	135	13.5	339	31.3
13	5	170	17	427	39.4
14	5	205	20.5	515	47.5
15	5	240	24	603	55.6

Fermentation data



pH



Ascaris eggs exposed to *n*-butyric acid and *n*-caproic acid in fecal sludge

